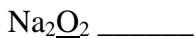
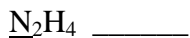
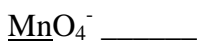
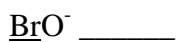
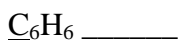
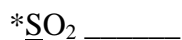


* VI-1 _____C _____NC
** VI-2 _____C _____NC

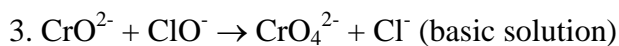
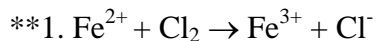
Chemistry 152
Worksheet 7

Name: _____

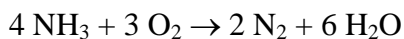
- A. (3.0 pts.) Give the oxidation number for the underlined atom in each of the following:



- B. (6.0 pts.) Balance each of the following reactions using the ion-electron method. Show all work.



- C. (1.0 pts.) Fill in the blanks for each of the following with respect to the reaction.



oxidizing agent _____

reducing agent _____